A: The previous version had a 32 bits crash when starting (on a 64 bits machine). In the new version (v6.9.0.4) it starts on a 64 bits machine. But it asks for the 32 bits one on a 32 bits machine : just ignore it. 32 bits version of the 32 bits client agent will start as well on 32 bits machine (also ask for the 32 bits client agent) This invention relates to an improved process for the preparation of hydrogen peroxide by the reaction of aqueous solutions of alkali or alkali metal salts of inorganic or organic acids with hydrogen peroxide in the presence of a catalyst and to a novel catalyst consisting of an iron and ruthenium compound. In the prior art, there are numerous catalysts for the preparation of hydrogen peroxide by the reaction of an aqueous alkali metal or alkali metal salt solution with hydrogen peroxide. These catalysts may be in the form of soluble inorganic substances such as silver, copper, zinc, lead, and the like. The most common forms of these catalysts are alkali metal or ammonium salts of the strong organic acids, such as nitric acid, acetic acid, phosphoric acid, formic acid and the like. More recently, catalysts which are organic in nature have been utilized and which are the salts of carboxylic acids, such as aldehydes and ketones, and the salts of sulfur acids, such as sulfonic acids, i.e., amine salts, sulfonium salts, and the like, and mixtures thereof. As an example of the state of the art, U.S. Pat. No. 3,702,788, by J. E. Anderson, discloses the preparation of hydrogen peroxide from alkali metal or alkali metal salts of organic acids and hydrogen peroxide by using carbonate salts of alkali metals or alkali metal salts of organic acids as the catalyst. The preferred catalyst is carbonic acid. It is noted, however, that when the process is used to prepare an aqueous solution of hydrogen peroxide, the resultant solution of hydrogen peroxide contains impurities such as acetic acid and formic acid. It is disclosed that these impurities will hydrolyze to form acid and organic salts which will give the solution of hydrogen peroxide a very unpleasant odor. This is particularly true when the solution of hydrogen peroxide is used in various industries

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ader.rar1. Field of the Invention The present invention relates to a dust removal filter device which removes dusts and other suspended or floating particulate materials in a gaseous or a liquid stream from the gaseous or liquid stream. More particularly, the present invention is directed to a dust removal filter device which is useful for removing dusts and other suspended or floating particulate materials from a

gas stream before the gas stream is exhausted to the atmosphere. 2. Description of 0cc13bf012

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